

# Periodic Table of the Elements

<b>1A</b> <b>1</b>	<b>Periodic Table of the Elements</b>																<b>8A</b> <b>18</b>
<b>1</b> <b>H</b> 1.008											<b>2</b> <b>He</b> 4.003						
<b>3</b> <b>Li</b> 6.941	<b>4</b> <b>Be</b> 9.012											<b>5</b> <b>B</b> 10.81	<b>6</b> <b>C</b> 12.01	<b>7</b> <b>N</b> 14.01	<b>8</b> <b>O</b> 16.00	<b>9</b> <b>F</b> 19.00	<b>10</b> <b>Ne</b> 20.18
<b>11</b> <b>Na</b> 22.99	<b>12</b> <b>Mg</b> 24.31	<b>3B</b> <b>3</b>	<b>4B</b> <b>4</b>	<b>5B</b> <b>5</b>	<b>6B</b> <b>6</b>	<b>7B</b> <b>7</b>	<b>8B</b> <b>8</b>	<b>8B</b> <b>9</b>	<b>8B</b> <b>10</b>	<b>1B</b> <b>11</b>	<b>2B</b> <b>12</b>	<b>13</b> <b>Al</b> 26.98	<b>14</b> <b>Si</b> 28.09	<b>15</b> <b>P</b> 30.97	<b>16</b> <b>S</b> 32.07	<b>17</b> <b>Cl</b> 35.45	<b>18</b> <b>Ar</b> 39.95
<b>19</b> <b>K</b> 39.10	<b>20</b> <b>Ca</b> 40.08	<b>21</b> <b>Sc</b> 44.96	<b>22</b> <b>Ti</b> 47.87	<b>23</b> <b>V</b> 50.94	<b>24</b> <b>Cr</b> 52.00	<b>25</b> <b>Mn</b> 54.94	<b>26</b> <b>Fe</b> 55.85	<b>27</b> <b>Co</b> 58.93	<b>28</b> <b>Ni</b> 58.69	<b>29</b> <b>Cu</b> 63.55	<b>30</b> <b>Zn</b> 65.38	<b>31</b> <b>Ga</b> 69.72	<b>32</b> <b>Ge</b> 72.64	<b>33</b> <b>As</b> 74.92	<b>34</b> <b>Se</b> 78.96	<b>35</b> <b>Br</b> 79.90	<b>36</b> <b>Kr</b> 83.80
<b>37</b> <b>Rb</b> 85.47	<b>38</b> <b>Sr</b> 87.62	<b>39</b> <b>Y</b> 88.91	<b>40</b> <b>Zr</b> 91.22	<b>41</b> <b>Nb</b> 92.91	<b>42</b> <b>Mo</b> 95.94	<b>43</b> <b>Tc</b> (98)	<b>44</b> <b>Ru</b> 101.07	<b>45</b> <b>Rh</b> 102.91	<b>46</b> <b>Pd</b> 106.42	<b>47</b> <b>Ag</b> 107.87	<b>48</b> <b>Cd</b> 112.41	<b>49</b> <b>In</b> 114.82	<b>50</b> <b>Sn</b> 118.71	<b>51</b> <b>Sb</b> 121.76	<b>52</b> <b>Te</b> 127.60	<b>53</b> <b>I</b> 126.90	<b>54</b> <b>Xe</b> 131.29
<b>55</b> <b>Cs</b> 132.91	<b>56</b> <b>Ba</b> 137.33	<b>57</b> <b>La</b> 138.91	<b>72</b> <b>Hf</b> 178.49	<b>73</b> <b>Ta</b> 180.95	<b>74</b> <b>W</b> 183.84	<b>75</b> <b>Re</b> 186.21	<b>76</b> <b>Os</b> 190.23	<b>77</b> <b>Ir</b> 192.22	<b>78</b> <b>Pt</b> 195.08	<b>79</b> <b>Au</b> 196.97	<b>80</b> <b>Hg</b> 200.59	<b>81</b> <b>Tl</b> 204.38	<b>82</b> <b>Pb</b> 207.20	<b>83</b> <b>Bi</b> 208.98	<b>84</b> <b>Po</b> (209)	<b>85</b> <b>At</b> (210)	<b>86</b> <b>Rn</b> (222)
<b>87</b> <b>Fr</b> (223)	<b>88</b> <b>Ra</b> (226)	<b>89</b> <b>Ac</b> (227)	<b>104</b> <b>Rf</b> (261)	<b>105</b> <b>Db</b> (262)	<b>106</b> <b>Sg</b> (266)	<b>107</b> <b>Bh</b> (264)	<b>108</b> <b>Hs</b> (277)	<b>109</b> <b>Mt</b> (268)	<b>110</b> <b>Ds</b> (281)	<b>111</b> <b>Rg</b> (281)	<b>112</b> <b>Cn</b> (285)	<b>113</b> <b>Nh</b> (286)	<b>114</b> <b>Fl</b> (289)	<b>115</b> <b>Mc</b> (289)	<b>116</b> <b>Lv</b> (293)	<b>117</b> <b>Ts</b> (293)	<b>118</b> <b>Og</b> (294)

<b>58</b> <b>Ce</b> 140.12	<b>59</b> <b>Pr</b> 140.91	<b>60</b> <b>Nd</b> 144.24	<b>61</b> <b>Pm</b> (145)	<b>62</b> <b>Sm</b> 150.36	<b>63</b> <b>Eu</b> 151.96	<b>64</b> <b>Gd</b> 157.25	<b>65</b> <b>Tb</b> 158.93	<b>66</b> <b>Dy</b> 162.50	<b>67</b> <b>Ho</b> 164.93	<b>68</b> <b>Er</b> 167.26	<b>69</b> <b>Tm</b> 168.93	<b>70</b> <b>Yb</b> 173.04	<b>71</b> <b>Lu</b> 174.97
<b>90</b> <b>Th</b> 232.04	<b>91</b> <b>Pa</b> 231.04	<b>92</b> <b>U</b> 238.03	<b>93</b> <b>Np</b> (237)	<b>94</b> <b>Pu</b> (244)	<b>95</b> <b>Am</b> (243)	<b>96</b> <b>Cm</b> (247)	<b>97</b> <b>Bk</b> (247)	<b>98</b> <b>Cf</b> (251)	<b>99</b> <b>Es</b> (252)	<b>100</b> <b>Fm</b> (257)	<b>101</b> <b>Md</b> (258)	<b>102</b> <b>No</b> (259)	<b>103</b> <b>Lr</b> (262)

## constants

$$R = 8.314 \text{ J/mol K}$$

$$R = 0.08206 \text{ L}\cdot\text{atm/mol}\cdot\text{K}$$

$$R = 62.36 \text{ L}\cdot\text{torr/mol}\cdot\text{K}$$

$$N_A = 6.022 \times 10^{23} / \text{mol}$$

$$h = 6.626 \times 10^{-34} \text{ J}\cdot\text{s}$$

Planck's constant

$$k = 1.38 \times 10^{-23} \text{ J/K}$$

Boltzmann constant

$$c = 3.00 \times 10^8 \text{ m/s}$$

$$F = 96485 \text{ C/mol } e^-$$

$$e = 1.602 \times 10^{-19} \text{ C}$$

charge on one electron

$$g = 9.81 \text{ m/s}^2$$

$$\mathcal{R} = 2.18 \times 10^{-18} \text{ J}$$

$$\mathcal{R} = 3.29 \times 10^{15} \text{ s}^{-1}$$

$$\mathcal{R} = 1.097 \times 10^7 \text{ m}^{-1}$$

Rydberg constants

## conversions

$$1 \text{ atm} = 760 \text{ torr}$$

$$1 \text{ atm} = 101325 \text{ Pa}$$

$$1 \text{ atm} = 1.01325 \text{ bar}$$

$$1 \text{ atm} = 14.7 \text{ psi}$$

$$1 \text{ bar} = 10^5 \text{ Pa}$$

$$1 \text{ L}\cdot\text{atm} = 101.325 \text{ J}$$

$$1 \text{ cal} = 4.184 \text{ J}$$

$$1 \text{ eV} = 1.602 \times 10^{-19} \text{ J}$$

$$1 \text{ W} = 1 \text{ J/s}$$

$$1 \text{ in} = 2.54 \text{ cm}$$

$$1 \text{ ft} = 12 \text{ in}$$

$$1 \text{ yd} = 3 \text{ ft}$$

$$1 \text{ mi} = 5280 \text{ ft}$$

$$1 \text{ \AA} = 10^{-10} \text{ m}$$

$$1 \text{ lb} = 453.6 \text{ g}$$

$$1 \text{ gal} = 3.785 \text{ L}$$

$$1 \text{ gal} = 231 \text{ in}^3$$

## H<sub>2</sub>O water data

$$C_{s,\text{ice}} = 2.09 \text{ J/g K}$$

$$C_{s,\text{water}} = 4.184 \text{ J/g K}$$

$$C_{s,\text{steam}} = 2.03 \text{ J/g K}$$

$$\rho_{\text{water}} = 1.000 \text{ g/mL}$$

$$\rho_{\text{ice}} = 0.9167 \text{ g/mL}$$

$$\Delta H_{\text{fus}} = 334 \text{ J/g}$$

$$\Delta H_{\text{vap}} = 2260 \text{ J/g}$$

Average Bond Energies (kJ/mol)					
C-H	413	C-C	347	C-Cl	339
C-O	358	C-N	305	C-Br	276
H-H	432	H-Cl	427	O-H	467
H-I	295	N-H	391	S-H	347
H-F	565	F-F	154	N-F	272
H-Br	363	Cl-Cl	239	Br-Br	193
I-I	149	C=O	799	C=C	614
O=O	495	N=N	418	C≡O	1072
C≡C	839	N≡N	941	C≡N	891